

# 课程大纲

## COURSE SYLLABUS

1.	<b>课程代码/名称</b> <b>Course Code/Title</b>	<b>Crystal Chemistry/晶体化学 MSE5017</b>
2.	<b>课程性质</b> <b>Compulsory/Elective</b>	专业选修课 (Specialty Elective Course)
3.	<b>课程学分/学时</b> <b>Course Credit/Hours</b>	3 学分/48 学时
4.	<b>授课语言</b> <b>Teaching Language</b>	English
5.	<b>授课教师</b> <b>Instructor(s)</b>	汪宏
6.	<b>是否面向本科生开放</b> <b>Open to undergraduates or not</b>	否
7.	<b>先修要求</b> <b>Pre-requisites</b>	无
8.	<b>教学目标</b> <b>Course Objectives</b>	
	<p>本课程主要讲述晶体、准晶体、非晶体的结构和对称性，以及金属、陶瓷、玻璃、聚合物等重要材料的晶体结构、原子模型和相关的晶体化学原理，使学生了解和掌握晶体物理和化学性质与晶体结构、微结构之间的联系，并理解有关原理和知识在材料设计制备和工程实践中的重要性。</p> <p>COURSE OBJECTIVES:</p> <ol style="list-style-type: none"> <li>1) To identify important raw materials and minerals as well as their names and chemical formulas.</li> <li>2) To describe the crystal structure of important materials and be able to build their atomic models.</li> <li>3) To learn the systematics of crystal and glass chemistry.</li> <li>4) To understand how physical and chemical properties are related to crystal structure and microstructure.</li> <li>5) To appreciate the engineering significance of these ideas and how they related to industrial products: past, present and future.</li> </ol>	
9.	<b>教学方法</b> <b>Teaching Methods</b>	
	<p>本课程将以基础知识讲授为主，结合前沿进展讲座，并穿插课程设计和实践的教学方法。教学方法不拘泥于课本和课件，综合基础知识、课程设计、前沿讲座，增加个性化交流机会，使学生理解和掌握晶体化学原理的基础知识和实际应用。</p> <p>This course will be focused on lecture fundamentals, combining the frontiers in crystal chemistry, and experimental skills. This course will not be based on teaching books, instead it will consist of lectures about basic knowledge, recent advances, and experimental skills. The lecturer will have personalized communications with students, to make sure that all the students have learned and grasped the fundamental concepts and application methods.</p>	
10.	<b>教学内容</b> <b>Course Contents</b>	
	<b>Section 1</b>	Introduction/ Chemical Elements, Minerals and Ceramic Raw Materials
	<b>Section 2</b>	Crystal Basic, Close Packing
	<b>Section 3</b>	Chemical Bonding and Electronegativity
	<b>Section 4</b>	Crystal Structure Symmetry: Crystal Systems and Bravais Lattices

<b>Section 5</b>	Crystal Structure Symmetry: Point Group
<b>Section 6</b>	Crystal Structure Symmetry: Space Group
<b>Section 7</b>	Crystal Chemistry Basic
<b>Section 8</b>	X-Ray and Diffraction
<b>Section 9</b>	Sizes of Atoms and Ions – Shannon-Prewitt Radii
<b>Section 10</b>	Pauling's Rules
<b>Section 11</b>	Typical Crystal Structures
<b>Section 12</b>	Crystal Field Theory
<b>Section 13</b>	Solid Solution
<b>Section 14</b>	Phase Diagram and Phase Transformation
<b>Section 15</b>	Defects in Crystals
<b>Section 16</b>	Glass and Quasi-crystal
<b>Section 17</b>	Newmann's Law and Tensor Properties
<b>Section 18</b>	Structure-Properties Relationship

**11. 课程考核**  
**Course Assessment**

考勤 (10%) + 作业 (25%) + 课程设计 (25%) + 期末考试 (40%)  
Attendance: 10% + Assignments: 25% + Project: 25% + Final: 40%

**12. 教材及其它参考资料**  
**Textbook and Supplementary Readings**

1. S. Trolrier-McKinstry, R. E. Newnham, Materials Engineering, Cambridge University Press, 2017, ISBN 978-1-107-10378-8
2. R. E. Newnham, Properties of Materials: Anisotropy, Symmetry, Structure. Oxford University Press, 2005, ISBN 0-19-852075-1
3. Edward R.T. Tiekink and Jagadese J. Vittal, Frontiers in Crystal Engineering. John Wiley & Sons, Ltd. 2006, ISBN: 0-470-02258-2
4. 钱逸泰, 结晶化学导论, 中国科学技术大学出版社, 2002, ISBN7-312-01088-1
5. R. C. Evans, An Introduction to Crystal Chemistry, Cambridge University Press, 1964